

CHEMICAL QUANTITIES STUDY GUIDE

Atomic Mass

- The mass of a single atom of an element (amu)
- Mass of one mole of the atom = atomic mass x amu
- $1 \text{ amu} = 1.66 \times 10^{-24} \text{ g}$

Avogadro's Number

- 6.022×10^{23}
- Six hundred 2 sextillion two hundred quintillion
- Represents the number of atoms that are in 1 mole of a substance

Molar Mass

- The sum of the total mass in grams of all the atoms that make up a mole of a particular molecule
- Molar mass of an element = number underneath name on periodic table
 - For a compound, calculate molar mass of each element and add

Mole Conversions

- Moles = mass/molar mass

$$\begin{array}{ccccc} \text{MASS (g)} & \begin{array}{c} \times \\ \hline \text{MOLAR MASS} \\ \hline + \end{array} & \boxed{\text{MOLE}} & \begin{array}{c} \times \\ \hline 6.022 \times 10^{23} \\ \hline + \end{array} & \text{FORMULA} \\ & & \begin{array}{c} \times \quad | \quad 22.4 \quad | \quad + \\ \hline \text{VOLUME} \end{array} & & \text{UNITS} \end{array}$$

Study Guide Chemical Quantities

Jill Meryl Levy



Study Guide Chemical Quantities:

Study Guide for Introductory Chemistry Iris Stovall, 1990 Student's Guide to Fundamentals of Chemistry Jo A. Beran, 2013-10-22 Student's Guide to Fundamentals of Chemistry Fourth Edition provides an introduction to the basic chemical principles. This book deals with various approaches to chemical principles and problem solving in chemistry. Organized into 25 chapters, this edition begins with an overview of how to define and recognize the more common names and symbols in chemistry. This text then discusses the historical development of the concept of atom as well as the historical determination of atomic weights for the elements. Other chapters consider how to calculate the molecular weight of a compound from its formula. This book discusses as well the characteristics of a photon in terms of its particle-like properties and defines the wavelength, frequency, and speed of light. The final chapter deals with the fundamental components of air and the classification of materials formed in natural waters. This book is a valuable resource for chemistry students, lecturers, and instructors. *Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice* Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 provides a complete guide to the fundamentals of chemistry. Page 4 of cover **Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual** Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18-chapter text designed to be used with Basics for Chemistry textbook. Each chapter contains Overview, Topical Outline, Skills, and Common Mistakes, which are all keyed to the textbook for easy cross-reference. The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms, a summary of general concepts, and a list of numerical exercises, while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook. The Fill-in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter, and the Skills section provides developed exercises to apply the new concepts in the chapter to particular

examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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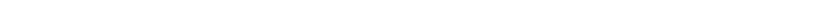
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