

Solubility and Concentration

1. Match the following. Write the correct letter in the blank provided.

1. Solute <u>B</u>	A. Tells you how much solvent is dissolved in the solution
2. Solvent <u>E</u>	B. The part that dissolves into the solution
3. Dissolve <u>D</u>	C. Increase solubility by moving molecules faster
4. Stirring <u>C</u>	D. When something is seen to disappear inside the solvent
5. Concentration <u>A</u>	E. The part that dissolves something

2. Determine if the following statements are true or false.

- Solubility increases if you stir a solute into a solution. F
- All solutes have the same solubility in a given solvent. F
- There is a limit on the amount of solute that can dissolve in a given solvent. T
- You can dissolve additional solutes in an unsaturated solution. T

3. Give three ways to make sugar dissolve faster. Heat Stirring and increasing surface area

4. Which one will dissolve faster? Circle the correct answer.

- Powdered sugar or granulated sugar?
- In hot water or in cold water?
- Stirred or not stirred?
- Large particles or small particles ?

5. 65 g of sugar is dissolved in 750 ml of water. What is the concentration of the solution?
0.08 g/ml

6. Which is more concentrated: 34 g of salt dissolved in 100 ml of water or 100 g of salt in 1500 ml of water?
34 g of Salt in 100ml of water

7. If the solubility of salt in water was determined to be 0.5 g/ml, would a solution with 50 g of salt in 150 ml of water be considered saturated?

no

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D Kirk



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Jacaranda Chemistry 1 VCE Units 1 and 2, learnON and Print Neale Taylor, Angela Stubbs, Robert Stokes, 2022-11-30 Developed by expert Victorian teachers for VCE students The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023 2027

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Elements of Physical Chemistry Peter William Atkins, Julio De Paula, 2017 This revision of the introductory textbook of physical chemistry has been designed to broaden its appeal particularly to students with an interest in biological applications

Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron s Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day

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Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity

Excel Revise in a Month TEE Chemistry Roy Skinner, 2003

Chemistry John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory

general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant

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