

Stoichiometry Study Guide

Students should be able to...

- Explain mole ratio
- Convert moles, grams, atoms/molecules
- Calculate theoretical yields when given the limiting reactant (Convert: moles A to moles B, grams A to moles B, moles A to grams B, grams A to grams B)
- Explain limiting reactant and excess.
- Calculate theoretical yields by finding the limiting reactant
- Calculate percent yield.
- Calculate percent composition.
- Calculate theoretical yield, limiting reactant, excess, and percent yield for a double replacement reaction (laboratory)

Practice Problems

Find the molar mass of each compound.

1. BaI_2
2. $\text{AgC}_2\text{H}_3\text{O}_2$

Convert.

3. 0.4 moles of Sn to atoms
4. 32 g of K_2O to molecules
5. 3.7 g of H_2 to moles
6. 800 g of CaCl_2 to moles
7. 0.75 moles of H_2O to grams
8. 9.0 g of Cl_2 to atoms
9. 0.5 moles of KNO_3 to grams

Find the percent composition for each element in the compound.

10. Mn_2O_7
11. AuNO_3

Stoichiometry



If you started with 3.7 moles of magnesium oxide, how many moles of magnesium hydroxide would be produced?



If 2.8 grams of sodium were used for the reaction, and an excess of iodine was present, how many moles of sodium iodide would be produced?



Stoichiometry Study Guide Answers

AW Chickering



Stoichiometry Study Guide Answers:

Whispering the Secrets of Language: An Mental Journey through **Stoichiometry Study Guide Answers**

In a digitally-driven earth where screens reign great and immediate transmission drowns out the subtleties of language, the profound techniques and emotional nuances concealed within words usually move unheard. Yet, nestled within the pages of **Stoichiometry Study Guide Answers** a interesting fictional treasure pulsating with organic emotions, lies an exceptional journey waiting to be undertaken. Written by a skilled wordsmith, this wonderful opus invites readers on an introspective journey, delicately unraveling the veiled truths and profound impact resonating within the very fabric of every word. Within the emotional depths of the emotional evaluation, we can embark upon a genuine exploration of the book is core subjects, dissect their charming writing design, and yield to the powerful resonance it evokes deep within the recesses of readers hearts.

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