

Lab Data

Freezing-point depression (°C)

Select

Freezing-point depression constant (°C/ <i>m</i>)		1.86		
Freezing-point of DI water (°C)		0.10		
	Solution 1	Solution 2	Solution 3	
Magnesium chloride	MgCl ₂	MgCl ₂	MgCl ₂	
Molar mass (g/mol)	95.21	95.21	95.21	
Mass of calorimeter (g)	17.22	17.45	17.33	
Volume of DI water (mL)	47.3	48.3	49.0	
Mass of magnesium chloride (g)	1.58	2.15	3.09	
Moles of magnesium chloride (mol)	0.0166	0.0226	0.0325	
Mass of calorimeter + solution (g)	66.12	67.78	69.53	
Mass of calorimeter + solution + ice (g)	110.45	113.96	114.54	
Freezing-point of solution (°C)	-1.12	-1.30	-1.96	
Mass of calorimeter + ice after decanting (g)	39.88	30.37	31.62	
Mass of ice added (g)	44.33	46.18	45.01	
Mass of ice remaining after decanting (g)	22.66	12.92	14.29	
Mass of melted ice (g)	21.67	33.26	30.72	
Mass of water in final solution (g)	68.99	81.44	79.83	
Mass of water in final solution (kg)	0.06899	0.08144	0.07983	
Molality of solution (<i>m</i>)	0.241	0.278	0.407	
Freezing-point depression (°C)	1.22	1.40	2.06	
van't Hoff factor				



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Lines:

RESE

Solutions Lab 33 Freezing Point Answers

Wilhelm Ostwald



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