Number

Single Replacement Reaction Worksheet

For each reaction use the activity series to complete the reaction. If no reaction will occur, write "no reaction". Make sure you balance your final answer.

- 1. Ag+KNO, No Reaction
- 2. Zn (NO3) → Zn (NO3) a +2Ag
- 3. Cl₂ 2K1 → 2 KCl + I2
- 4. Cu + FeSO. → NO Reaction
- 52 Fe +3Pb(NOs)2 2 Fe (NOs)3 + 3 Pb
- 6. Cu + Ala(SOL)) No Reaction
- 7.2AI BPB(NO))= 2 A I (NO) 3+3Pb
- 8. Clo 2Nal → 2 lb c/ + I2
- 4. Fe BABC:HO2-2 3 Ag + Fr (C3H3C2);
- 102A1 3C0Cb 3 Ca + 2A1 C13
- 11. Br2 + Cal2 → CaBr2 + I2
- 12) Fe 3 CusO, 3 Cu + Fp 2 (504)3
- 13. Cl2 + Mgl2 Hg Cl2 + I2

| Activity Series | | Moss |
|-----------------|-----------------|------|
| Michala | Nonmetals | 1000 |
| Lit | F ₂ | 1 1 |
| Rb | Cl ₂ | 1 1 |
| K | Br ₂ | |
| Cis | I ₂ | |
| Ba | | |
| Sir | | 1 1 |
| Ca | | 1 1 |
| Nin. | | |
| Pv4fdz | | 1 1 |
| AL | | 1 1 |
| Ti. | | |
| Num | | |
| Zm | | |
| Cir | | |
| Fe | | |
| Co | | 1 1 |
| Ni | | 1 1 |
| Sm | | 1 1 |
| Pb | | 1 1 |
| H ₂ | | 1 1 |
| Cu | | |
| Ag | | |
| Ani | | |

Least

Single Replacement Reaction Ws Key

Yijin Wang

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