

Exercises in Nuclear Reactor Analysis

Solutions to the exercises of the text book "Nuclear
Reactor Analysis" by James J. Duderstadt and Louis J.
Hamilton

Prepared by W.F.G. van Rooijen

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Ian Pickup



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web the pcl 3 lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note the lone pair dots without bonds on top of p just like for n in the previous example for nh 3 chlorine cl

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