

AP Chemistry Final Study Guide

Unit 1: Atomic Structure and Properties (7%–9%)

- At STP, one mole of gas always occupies 22.4 L
- **Mass Spectroscopy:**
 - Atoms have different isotopes (carbon-12, carbon-14, etc) and this will affect the average atomic mass
 - The most abundant isotope will have the greatest effect
- **Coulomb's Law:**
 - The greater the charge of the nucleus, the more energy is required to ionize an electron
 - The closer an electron is to the nucleus, the more energy is required to ionize an electron
- **Electromagnetic Radiation:**
 - When atoms absorb energy, electrons jump to higher energy levels
 - When electrons drop from higher to lower energy levels, atoms give off electromagnetic radiation energy
- **Kinetic Energy of Ionized Electrons:**
 - The faster an ejected electron is going, the more kinetic energy it has
 - Electrons originally further away from the nucleus require less energy to eject, so it will have more kinetic energy
- **Electron Configuration:**
 - When entering the d-block (transition metals), the energy level drops by 1 unless you are using shorthand notation
 - Nickel: $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 4s^2$ (notice the d-block)
 - Nickel (with shorthand): $[\text{Ar}] 4s^2 3d^8$
- **Law of Definite Proportions:**
 - A pure chemical compound broken down into elements always contains elements of a fixed ratio, independent of where and how it was created
- **Law of Multiple Proportions:**
 - When elements form compounds, the proportions of the elements in those chemical compounds can be expressed in small whole number ratios
- **Aufbau Principle:**
 - When building up the electron configuration of an atom, electrons are placed in orbitals, subshells, and shells in order of increasing energy

Spring Semester Environmental Chemistry Final Study Guide

Institute for European Environmental Policy, Commission of the European Communities. Directorate-General Environment, Nuclear Safety and Civil Protection

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Resources in Education, 1998-05 **EC Study Guide to Environment-related Courses** Institute for European Environmental Policy, Commission of the European Communities. Directorate-General Environment, Nuclear Safety and Civil Protection, 1993 **Annual Catalogue** United States Air Force Academy, 1984 **Inventory of Federal Energy-related Environment and Safety Research for ...**, 1980 **Inventory of Federal Energy-related Environment and Safety Research for FY 1979**, 1980 **Bulletin of Information** United States Coast Guard Academy, 1982 *Research in Education*, 1974 **Selected Water Resources Abstracts**, 1989 **Guide to American Graduate Schools** Harold Doughty, 1997 Anyone faced with the overwhelming selection of graduate schools knows how valuable clear and concise information can be And that's just what they'll get with this comprehensive guide now fully updated and in its eighth edition Every aspect of graduate study is covered including enrollments locations and housing situations for more than 1 200 accredited institutions fields of study offered by each institution admissions and degree requirements financial aid and grant opportunities details on scholarships fellowships assistantships and internships Organized in an easy to use format that allows for comparison among the schools and their programs this guide is the perfect source to begin planning for higher education *Curriculum Handbook with General Information Concerning ... for the United States Air Force Academy* United States Air Force Academy, 2002

The book delves into Spring Semester Environmental Chemistry Final Study Guide. Spring Semester Environmental Chemistry Final Study Guide is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Spring Semester Environmental Chemistry Final Study Guide, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
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 - Chapter 3: Spring Semester Environmental Chemistry Final Study Guide in Everyday Life
 - Chapter 4: Spring Semester Environmental Chemistry Final Study Guide in Specific Contexts
 - Chapter 5: Conclusion
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- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Spring Semester Environmental Chemistry Final Study Guide.

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