

AUTOMATED CALENDAR 21-DAY HABIT-BUILDING PLANNER

21 Days										habit formation 2023 Year 5 moon 25 day			
Monday	Tuesday	Wednesday	Thursday	Friday	Saturday	Sunday	habit tracker			Notes			
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web a nucleus changes into a new element by emitting nuclear radiations these changes are described using nuclear equations example ${}^{86}_{219}\text{textrm{rn}} \rightarrow {}^{84}_{215}\text{textrm{po}}$

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web may 30 2014 answer link nuclear equations can be solved quite simply let's do one example of alpha decay and one of beta decay first a quick revision of radioactive decay during alpha decay an alpha particle is emitted from the nucleus it is the equivalent of a helium atom i.e. it has a mass of 4 and an atomic number of 2

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web nuclear reactions can be represented by balanced equations of nuclei in the A_ZX form A_ZX notation for atomic nuclei the top number A represents the nucleon number or the mass number nucleon number a total number of protons and neutrons in the nucleus the lower number Z represents the proton or atomic number

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web the carbon 14 nucleus is unstable it's radioactive it's going to undergo spontaneous decay it's going to try to get a better neutron to proton ratio so let's look at the nuclear equation which represents the spontaneous decay of carbon 14 so here is

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web sep 12 2022 the half life of the nuclei and thus the material is $t_{1/2} = 693$ million years 35 a the decay constant is $\lambda = 1.99 \times 10^{-5} \text{ s}^{-1}$ b since strontium 91 has an atomic mass of 90.90 g

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