

Solution Manual

OXFORD

Molecular Quantum Mechanics

Fifth edition

Peter Atkins and Ronald Friedman

$$\frac{\partial}{\partial x} \Psi = ik_x \Psi$$

$$\frac{\partial^2}{\partial x^2} \Psi = -k_x^2 \Psi$$

$$E = \frac{1}{2} \hbar^2 k^2 = \hbar^2 \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} \right) \Psi$$

$$\nabla^2 \Psi + V \Psi = E \Psi$$



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