# Determination of the Solubility-Product Constant for a Sparingly Soluble Salt

A. Preparation	of a	Call	bratic	m Curve
Initial [CrO <sub>4</sub> <sup>2</sup> ]	0.0	024	M	

Volume of 0.0024 M K <sub>2</sub> CrO <sub>4</sub>	Total volume	[0:0,3-]	Absorbance
1_1.00 mL	100. mL	1	0.060
2 5.00 mL	100. mL		0.299
a 10.00 mL	100. mL		0.600
4 15.00 mL	100. mL		0.840
And a boundary of the contract	and the second second second	THE RESERVE OF THE PARTY OF THE	No. of Concession, Name of Street, or other party of the last of t

Average motar absorption coefficient

Standard deviation (show calculations)

# B. Determination of the Solubility-Product Constant

	Absorbesor	[CiO <sub>2</sub> <sup>2</sup> ]	[Ag*]	Kee of Ar-CrO.
380	0.720		110000	
2	0.750	Marie Contract of the Contract		
3.	0.780			

33-W...gog

# **Solubility Product Lab Answers**

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