

Using a No. 2 pencil, print and block out: *Your lab section (such as 062) and *Your name, and Your ID#. Sign your name where it says SIGNATURE.

Part I: 30 Multiple Choice (4 pts each; total 120) Part II: 2 Written (48 pts) Total pts = 168

Part I. Multiple Choice Choose the best answer. No penalty for guessing.

1. In which of the groups below do boiling points increase from left to right (lowest to highest)?
Diethyl ketone, hexane, and 1-pentanol have similar molar masses (86 to 88).

- a) diethyl ketone, 1-pentanol, hexane
b) diethyl ketone, hexane, 1-pentanol
c) hexane, diethyl ketone, 1-pentanol
d) 1-pentanol, diethyl ketone, hexane

2. Which of these compounds are constitutional isomers of C_6H_{14} ?



- a) I and II b) I and III c) II and III d) II and IV e) III and IV

3. Esters are synthesized from two classes of organic compounds. These two types of compounds are:

- a) carboxylic acids and bases b) amines and alcohols
c) alcohols and carboxylic acids d) amines and alkenes e) alkenes and bases

4. Since hemiacetals are unstable, they are seldom found in significant quantities in reaction mixtures.

An exception to this statement is the formation of hemiacetals _____.

- a) in an acid-catalyzed reaction
b) in a base-catalyzed reaction
c) from aldehyde and alcohol groups within the same molecule
d) from carbonyl compounds and alcohols with different R groups
e) from carbonyl compounds and alcohols with identical R groups

5. Which of the following molecules exhibits cis/trans isomerism?

- a) $CH_2=CHCH_3$ b) $CH_3CH=CHCH_3$
c) $(CH_3)_2C=CHCH_2CH_3$ d) $CH_3CH_2CH=CH_2$

4, 27, 6

Spring 2013 Chemistry Final Exam Answers

U A Th Brinkman, J. G. Dorsey, J R Durig

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