#### Worksheet: Solutions Introduction

Name\_\_\_\_

Explain why solutions are classified as mixtures instead of compounds.

 Solutions are \_\_\_\_\_ mixtures made up of very small particles that are actually molecules, \_\_\_\_\_, or \_\_\_\_\_.

 Solutions are said to be in a \_\_\_\_\_\_ phase even though the components may have been in different phases before the solution was formed.

4. Pure gold is 24 carat. 14-carat gold contains 14 parts gold and 10 parts other metals. 14-carat gold is said to be a(n) \_\_\_\_\_\_, which is a type of \_\_\_\_\_\_, solution. An example of a gaseous solution is \_\_\_\_\_, which is made up mostly of \_\_\_\_\_\_ and nitrogen when dry. The most common solutions are \_\_\_\_\_\_ solutions.

Define miscible:

Define immiscible:

 Because of the \_\_\_\_\_\_\_ you can see the light beams from car headlights in a fog.

Multiple choice: To increase the rate of solution of a solid in water,

a. increase the pressure over the water.

b. decrease the pressure over the water.

c. crush the particles of the solid.

d. chill the water.

 If 15 grams of iodine are dissolved in 1000 mL of alcohol, the alcohol is the (solute, solvent) and the solution is said to be a(n) \_\_\_\_\_\_\_.

 A substance that dissolves other materials is a (solute, solvent). The substance being dissolved is a (solute, solvent).

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Matthew J. Johll

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