

Chapter 19 Study Guide: Acids, Bases & Titration – Answers

1. Differentiate between Arrhenius, Brønsted-Lowry & Lewis acids and bases.

	ACIDS	BASES
Arrhenius	Form hydronium ions (H_3O^+) in aqueous sol'n	Form hydroxide ions (OH^-) in aqueous sol'n
Brønsted-Lowry	Proton (H^+) donors	Proton (H^+) acceptors
Lewis	Electron pair acceptors	Electron pair donors

2. Label the acid (A), base (B), conjugate acid (CA) and conjugate base (CB) in each of the following reactions:



3. Write the names or formulas for the following acids and bases:

- a. H_2S – hydrosulfuric acid
 b. $\text{HC}_2\text{H}_3\text{O}_2$ – acetic acid
 c. NaOH – sodium hydroxide
 d. HNO_3 – nitric acid
 e. Nitrous acid – HNO_2
 f. Calcium hydroxide – $\text{Ca}(\text{OH})_2$
 g. hydrochloric acid – HCl

Use the following formulas for the remaining questions: (these will not be given to you on the test!)

$$K_w = [\text{H}_3\text{O}^+][\text{OH}^-] = 1.0 \times 10^{-14}$$

$$\text{pH} = -\log[\text{H}_3\text{O}^+]$$

$$\text{pOH} = -\log[\text{OH}^-]$$

$$\text{pH} + \text{pOH} = 14$$

$$M_A \cdot V_A \cdot n_A = M_B \cdot V_B \cdot n_B$$

4. What is the pH of
- 1.34×10^{-4}
- M solution of hydrochloric acid?

$$\text{pH} = -\log [1.34 \times 10^{-4}\text{M}] = 3.873$$

5. Determine the pH and pOH of a
- 1.85×10^{-6}
- M NaOH solution.

$$\text{pOH} = -\log [1.85 \times 10^{-6}\text{M}] = 5.733$$

$$\text{pH} = 14 - 5.73 = 8.267$$

6. What is the
- $[\text{OH}^-]$
- of an aqueous solution that has a pH of 5.0?

$$\text{pOH} = 9 \quad [\text{OH}^-] = 1 \times 10^{-9} \text{ M}$$

7. What is the pH and the pOH of a solution with a volume of 2000. mL containing 35 grams of NaOH? Is this solution acidic or basic?

$$\frac{35\text{gNaOH}}{2.000\text{L}} \times \frac{1\text{molNaOH}}{40.00\text{gNaOH}} = 0.44\text{M NaOH}$$

$$\text{pOH} = -\log[0.44\text{M}] = 0.36 \quad \text{pH} = 14 - 0.36 = 13.64 \quad \text{basic}$$

Study Guide Acids Bases And Titration

Padhraic Smyth



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Table of Contents Study Guide Acids Bases And Titration

1. Understanding the eBook Study Guide Acids Bases And Titration
 - The Rise of Digital Reading Study Guide Acids Bases And Titration
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide Acids Bases And Titration
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide Acids Bases And Titration
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide Acids Bases And Titration
 - Personalized Recommendations
 - Study Guide Acids Bases And Titration User Reviews and Ratings
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 - Interactive Elements Study Guide Acids Bases And Titration
- 8. Staying Engaged with Study Guide Acids Bases And Titration
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide Acids Bases And Titration
- 9. Balancing eBooks and Physical Books Study Guide Acids Bases And Titration
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide Acids Bases And Titration
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Study Guide Acids Bases And Titration
 - Setting Reading Goals Study Guide Acids Bases And Titration
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Study Guide Acids Bases And Titration
 - Fact-Checking eBook Content of Study Guide Acids Bases And Titration
 - Distinguishing Credible Sources
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 - Exploring Educational eBooks
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 - Interactive and Gamified eBooks

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web a score of 18 is required for the national diploma extended curriculum assessment procedures applicants with a inal aps of 22 and more will be admitted to the programme applicants with a score of 20 to 21 will be admitted to

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web 1 registration the first thing you need to do is to register with the professional body that s cima acca cfa or icb with nated you can skip this step 2 exemptions if you have an existing qualification you ll want to apply for exemptions with cima or acca the other institutes don t allow for exemptions 3 expert support

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web the national diploma in management nqf level 6 allows the learner to work towards a nationally recognised whole qualification the qualification will allow both those in formal education and those already employed in organisations access due to its flexibility

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web with a score of 20 to 21 will be admitted to the national diploma extended curriculum applicants with a score of 18 to 19 will be required to do the tut potential assessment for applicants with a national certificate vocational admission requirement s a national certificate vocational with at least 50 for english home

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web feb 14 2023 a national n diploma is not equivalent to a national diploma to enter a national n diploma programme you need a matric certificate or a completed nated level 3 qualification the national n diploma qualifications are focused on specific trades like engineering hair care financial management motor mechanic and so on

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