Student Exploration: Stoichiometry

Vocabulary: Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, molar mass, mole, molecular mass, stoichiometry

	Get the Gizmo ready:	
Activity A:	 Check that the equation is still: 	
Moles	Fe ₂ O ₃ + 3CO 2Fe + 3CO ₂	144
	 If not, click New equation until it reappears. 	m.

Question: How do scientists find the molecular mass and molar mass of a substance?

 <u>Calculate</u>: The molecular mass of a molecule is the sum of the masses of each atom in the molecule. The unit of molecular mass is the universal mass unit (u).

Iron's atomic mass is 55.85 u, carbon's mass is 12.01 u, and oxygen's mass is 16.00 u.

- Calculate the molecular mass of carbon monoxide (CO) by adding the atomic mass of carbon and the atomic mass of oxygen:
- B. Calculate the molecular mass of carbon dioxide (CO₂): 44.01g
- Infer: A mole of a substance has a mass in grams that is equal to the molecular mass. For example, a carbon atom has a mass of 12.01 u. A mole of carbon has a mass of 12.01 g. Based on their molecular masses, list the molar mass of each substance:

Check your answers on the Gizmo by inspecting the middle row of tiles on the right side of the Gizmo. These tiles show the units "1 mol" on top and "g" below.

Practice: Hydrogen has an atomic mass of 1.01 u. What is the molar mass of these substances?

Student Exploration Stoichiometry Answer

Javier García-Martínez, Elena Serrano-Torregrosa

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Improvina Student Comprehension of Stoichiometric Concepts Connie Lynn Bannick Kemner, 2007 Chemistry Education Javier García-Martínez, Elena Serrano-Torregrosa, 2015-05-04 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future Adopting a practice oriented approach the current challenges and opportunities posed by chemistry education are critically discussed highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them The main topics discussed include best practices project based education blended learning and the role of technology including e learning and science visualization Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively from experience chemistry professors to secondary school teachers from educators with no formal training in didactics to frustrated chemistry students U-M Computing News ,1990 ENC Focus, 2000 Innovative Curriculum Materials ,1999 Conference Proceedings. New Perspectives in Science Education Pixel, 2017 Writing and Learning in the Science Classroom Carolyn S. Wallace, Brian B. Hand, Vaughan Prain, 2004-03-31 This volume is of interest to science educators graduate students and classroom teachers. The book will also be an important addition to any scholarly library focusing on science education science literacy and writing This book is unique in that it synthesizes the research of the three leading researchers in the field of writing to learn science Carolyn S Wallace Brian Hand and Vaughan Prain It includes a comprehensive review of salient literature in the field detailed reports of the authors own research studies and current and future issues on writing in science The book is the first to definitely answer the guestion Does writing improve science learning Further it provides evidence for some of the mechanisms through which learning occurs It combines both theory and practice in a unique way Although primarily a tool for research classroom teachers will also find many practical suggestions for using writing in the science classroom Resources in Education ,1995 Space Research ,2003

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