

Experiment 35

Advance Study Assignment: Spot Tests for Some Common Anions

1. Each of the observations in the following list was made on a different solution. Given the observations, state which ion studied in this experiment is present. If the test is not definitive, indicate that with a question mark.
 - A. Addition of 6 M NaOH and Al to the solution produces a vapor that turns red litmus blue.
Ion present:
 - B. Addition of 6 M HCl produces a vapor with an acrid odor.
Ion present:
 - C. Addition of 6 M HCl produces an effervescence.
Ion present:
 - D. Addition of 6 M HNO₃ plus 1.0 M BaCl₂ produces a precipitate.
Ion present:
 - E. Addition of 6 M HNO₃ plus 0.10 M AgNO₃ produces a precipitate.
Ion present:
 - F. Addition of 6 M HNO₃ plus 0.5 M (NH₄)₂MoO₄ produces a precipitate.
Ion present:
2. An unknown containing one or more of the ions studied in this experiment has the following properties.
 - A. No effect is observed on addition of 6 M HNO₃.
 - B. A white precipitate is observed on addition of 1.0 M BaCl₂ to the solution from Part A.
 - C. No effect is observed on addition of 0.10 M AgNO₃ to the solution from Part A.
 - D. A yellow precipitate is observed on addition of (NH₄)₂MoO₄ to the solution from Part A.

On the basis of this information, which ions are present, which are absent, and which are in doubt?

Present

Absent

In doubt

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LP Steffe



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